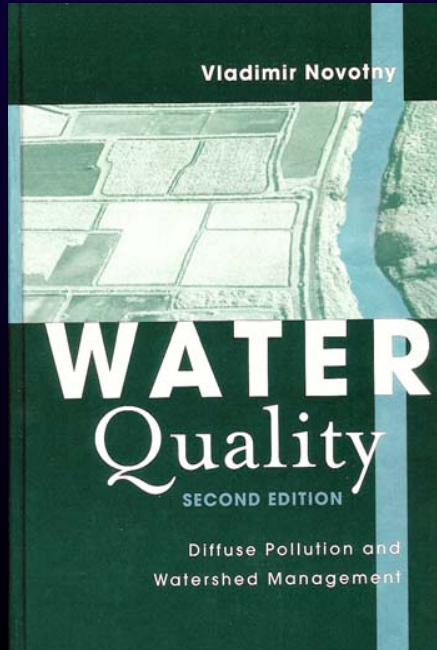


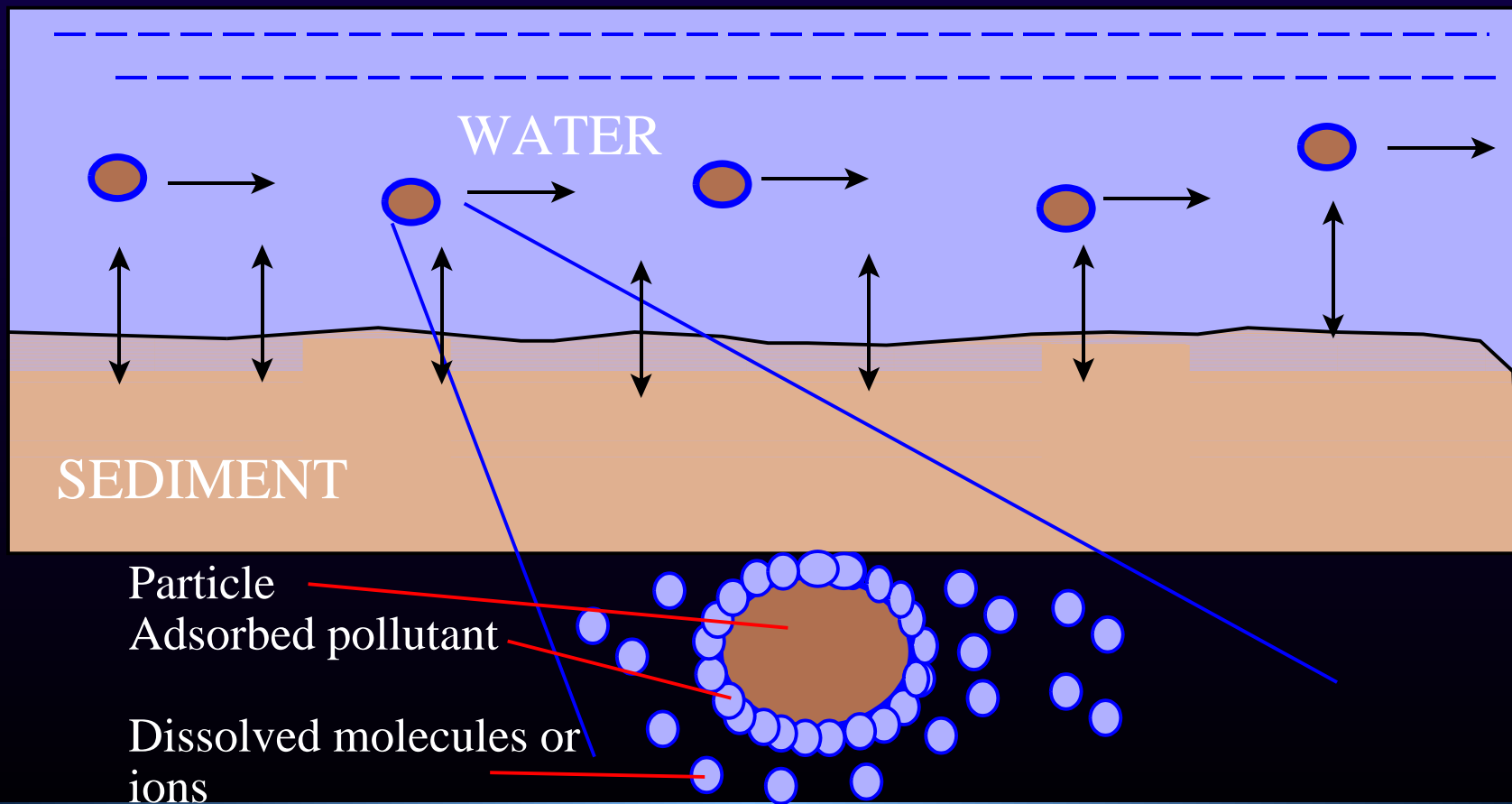
MODELING TOXIC COMPOUNDS



Supplement to Chapter 13

© Vladimir Novotny

TWO PHASE MOVEMENT OF TOXIC COMPOUNDS



PARTITIONING CONCEPT

Pollutants (hydrophobic organic toxic compounds, ammonium, phosphorus, toxic metals) exist in water or in sediment as

DISSOLVED ← **ADSORBED**
Ionic form Precipitated

Equilibrium

Pb^{++} , Cu^{++}

CuS , PbS

In particulate form most of the contaminant are not toxic. The dissolved (ionic form) is toxic to aquatic organisms

The adsorption equilibrium is affected by a number of ligands such as pH, organic particulate matter, redox potential (O_2 concentration), clay content, chlorides (salinity), etc.

The adsorption equilibrium between dissolved (ionic) and adsorbed (particulate) forms is expressed by adsorption isotherms , e.g,

Langmuir

$$r = \frac{Q^0 b C_e}{1 + b C_e}$$

Linear

$$r = \Pi C_e$$

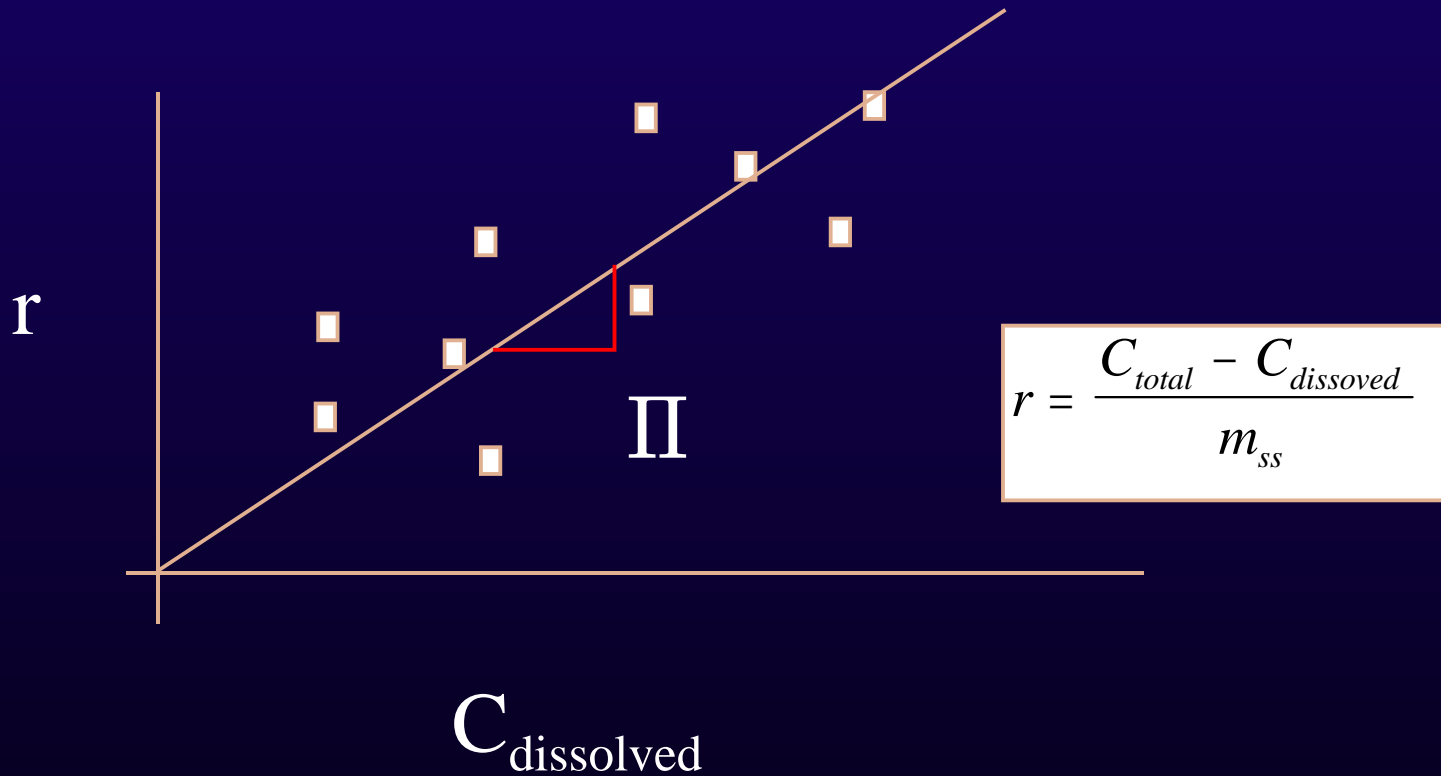
r = adsorbed concentration of the contaminant ($\mu\text{g/g}$ of sediment)

C_e = dissolved (ionized) concentration of the contaminant ($\mu\text{g/L}$)

Q^0 = adsorption maximum at fixed temperature ($\mu\text{g/g}$ of sediment)

b = a constant related to energy of the net enthalpy of adsorption ($\text{L}/\mu\text{g}$)

Π = linear partitioning coefficient (L/g)



Linear Adsorption Isotherm

m_{ss} = concentration of suspended solids or
particulate carbon (g/L)

Total pollutant concentration in water

$$C_{\text{total}} = C_d + C_p = C_d + m_{\text{ss}} r = C_d (1 + \Pi m_{\text{ss}})$$

In water

$$C_e = \frac{C_{\text{total}}}{1 + \Pi m_{\text{ss}}}$$

In sediment

$$C_e = \frac{C_{\text{total}}}{\theta + \Pi m_{\text{ss}}}$$

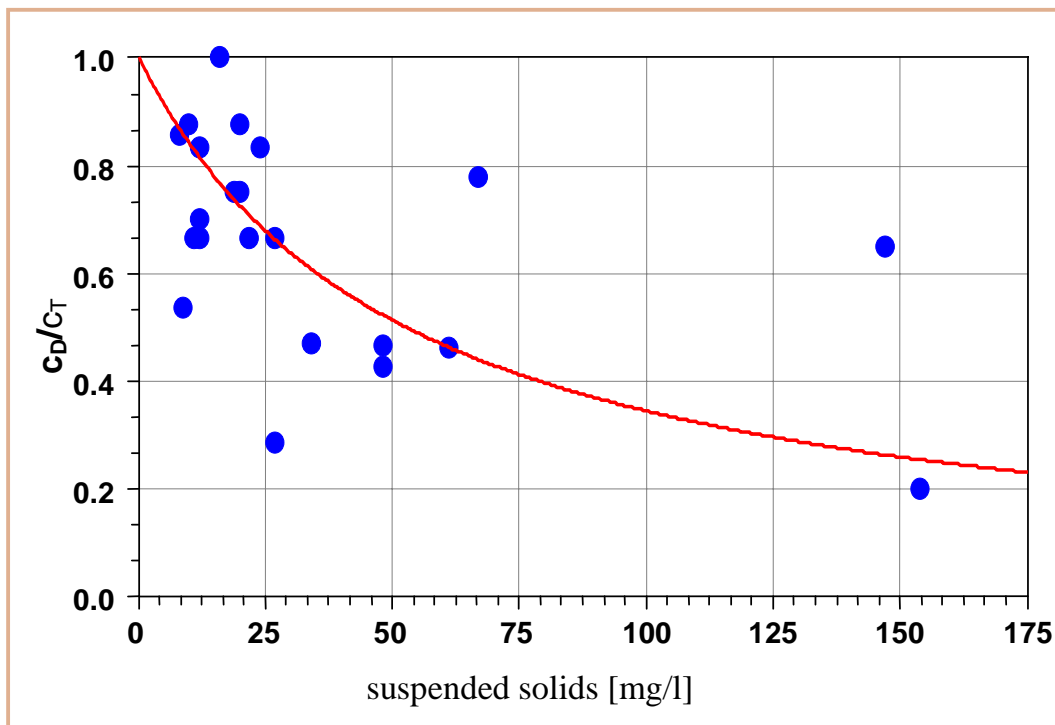
$\theta = \text{porosity}$

PARTITIONING

Magnitude of the partitioning coefficient ranges from 1 L/g to 10^6 L/g

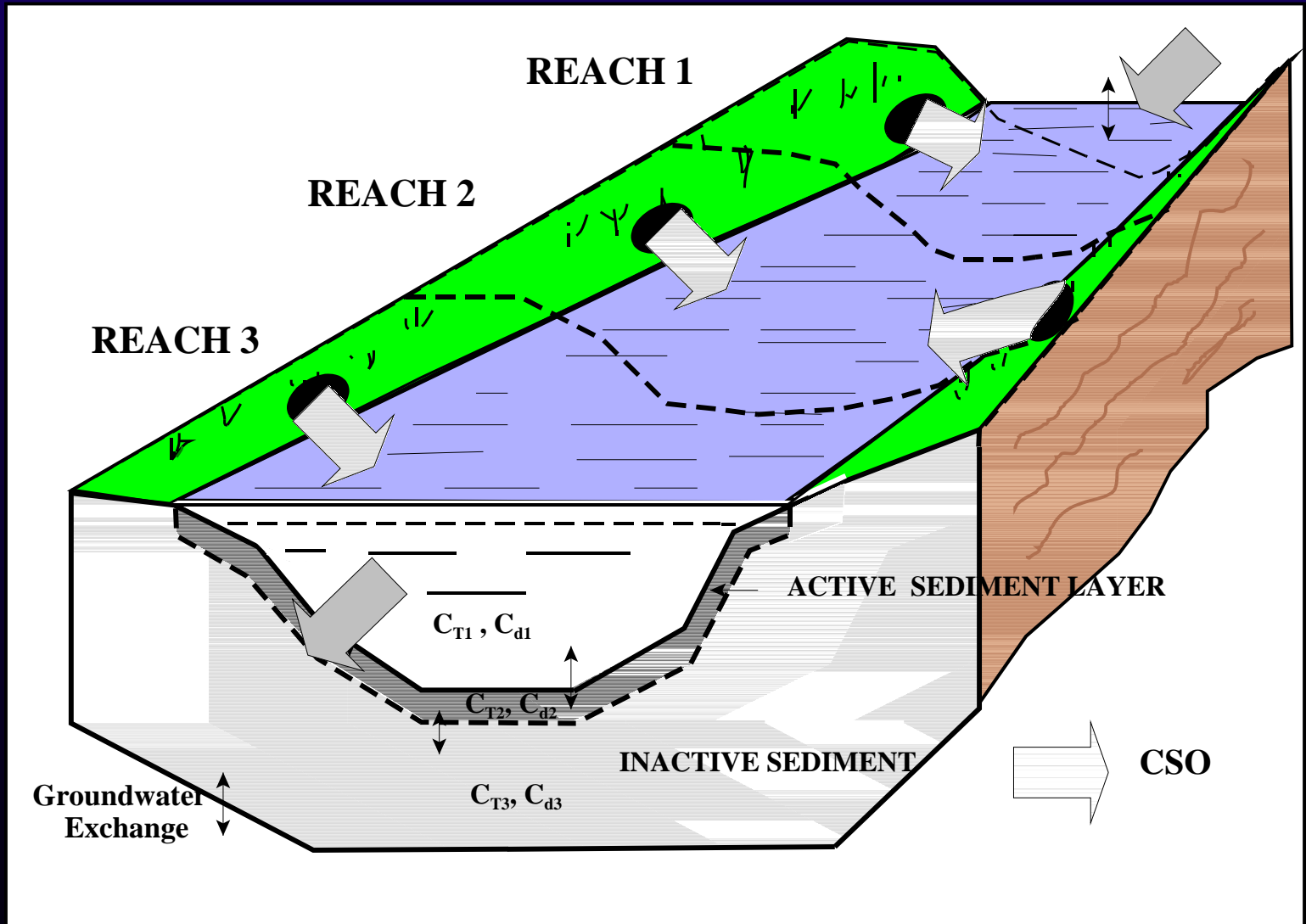
Solvents

PCBs, DDT



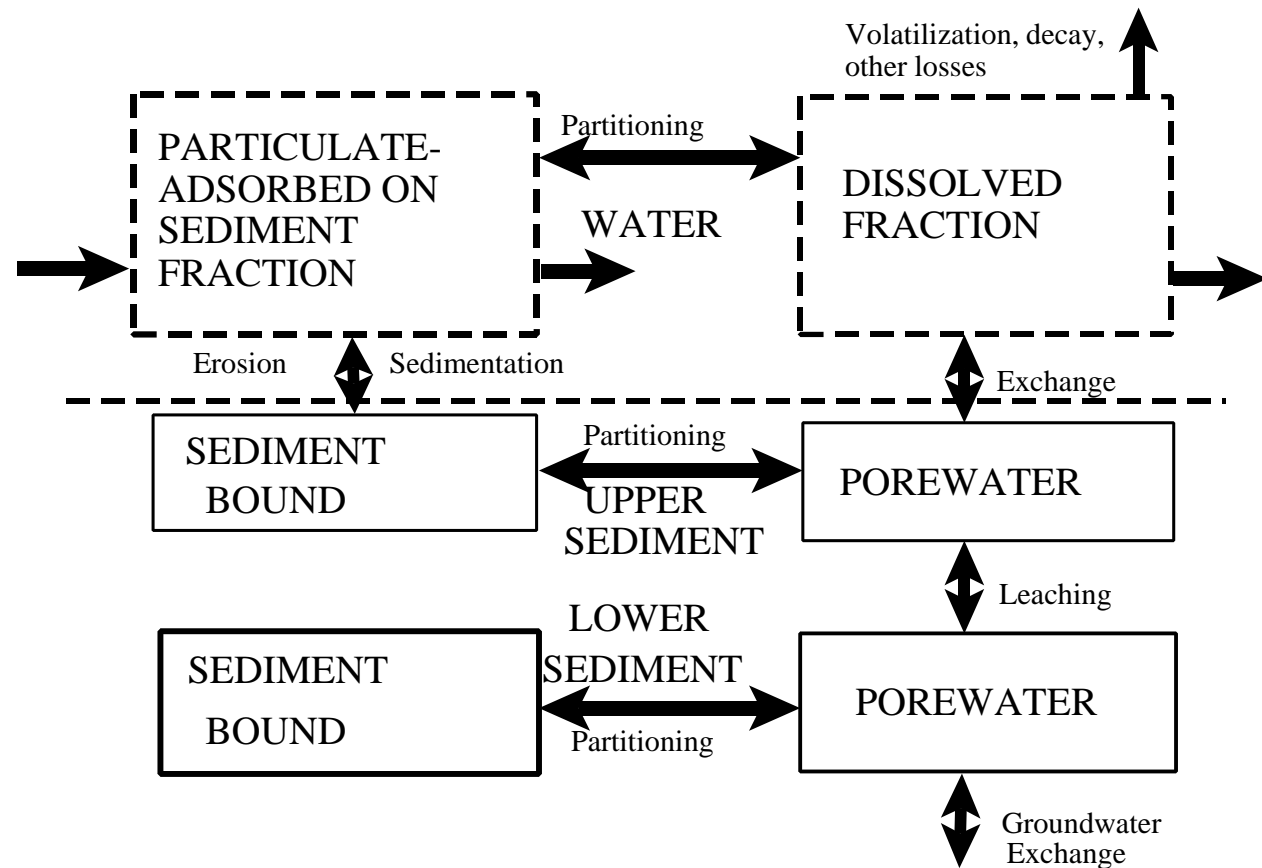
For toxic organics Π is related to the octanol partitioning coefficient K_{ow}

North Avenue Dam in Milwaukee



SCHEMATIC OF THE TOXIC MODEL

SCHEMATIC OF THREE LAYER MODEL



TWO PHASE TOXIC COMPOUND TRANSFER

IN
WATER

Dissolved

$$\frac{dC_d}{dt} = \frac{W_d}{V} - \frac{C_d}{t_0} - (K_1 + K_c + K_a)C_d + K_2C_p$$

Particulate

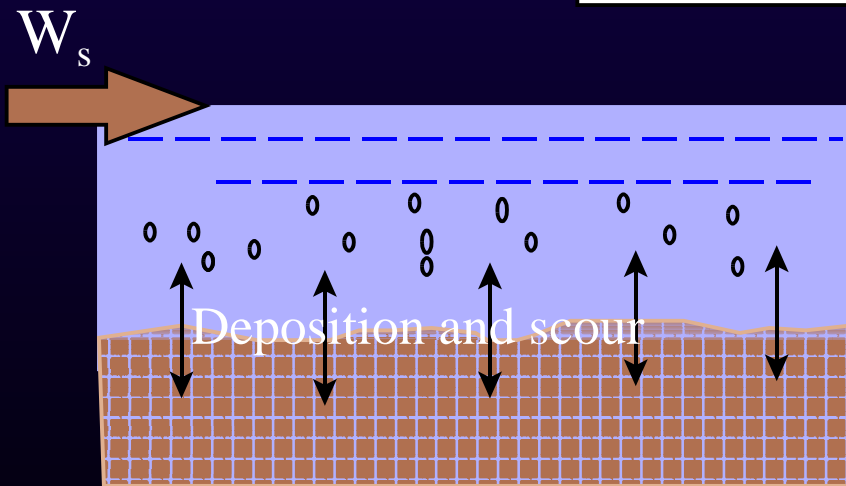
$$\frac{dC_p}{dt} = \frac{W_p}{V} - \frac{C_p}{t_0} + K_1C_d - (K_2 + K_p + K_s)C_p$$

C_d =dissolved concentration;
 V = volume of the computational segment
 t_0 =detention time within the segment, V/Q
 K_2 = desorption rate coefficient
 K_d =transfer (volatilization) rate coeff.
 $K_s=v_s/H$ =sediment transfer rate
 v_s =settling (resuspension) velocity

C_p =adsorbed fraction concentration
 Q =flow through the segment
 K_1 =adsorption rate coefficient
 K_c = decay rate coefficient, dissolved
 K_p =decay rate coeff., particular
 W_c, W_p = mass input of the dissolved and adsorbed fractions

SEDIMENT MASS BALANCE

$$\frac{dm}{dt} = \frac{m_i}{t_0} - m \left[\frac{1}{t_0} + K_s \right]$$



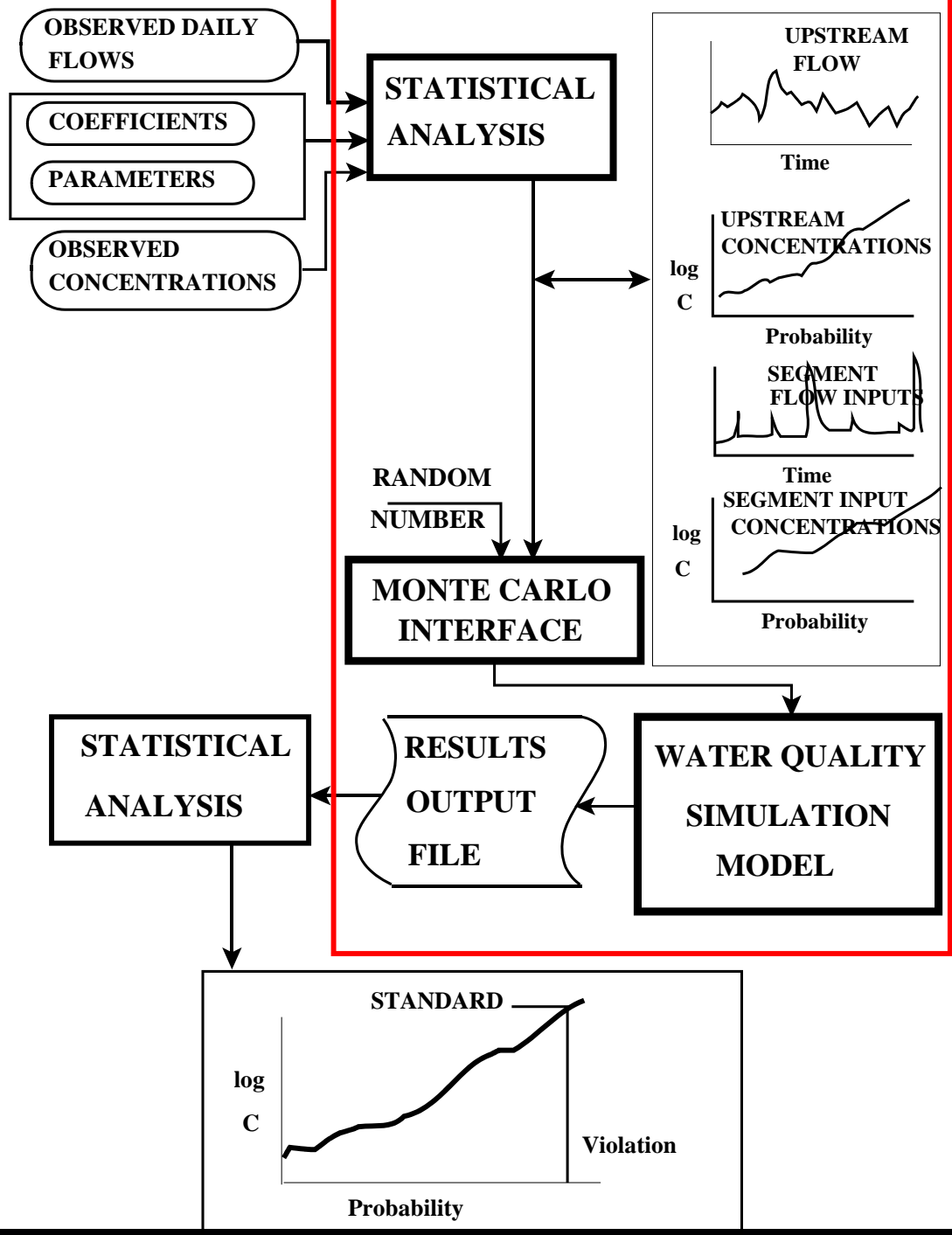
m = concentration of suspended sediment

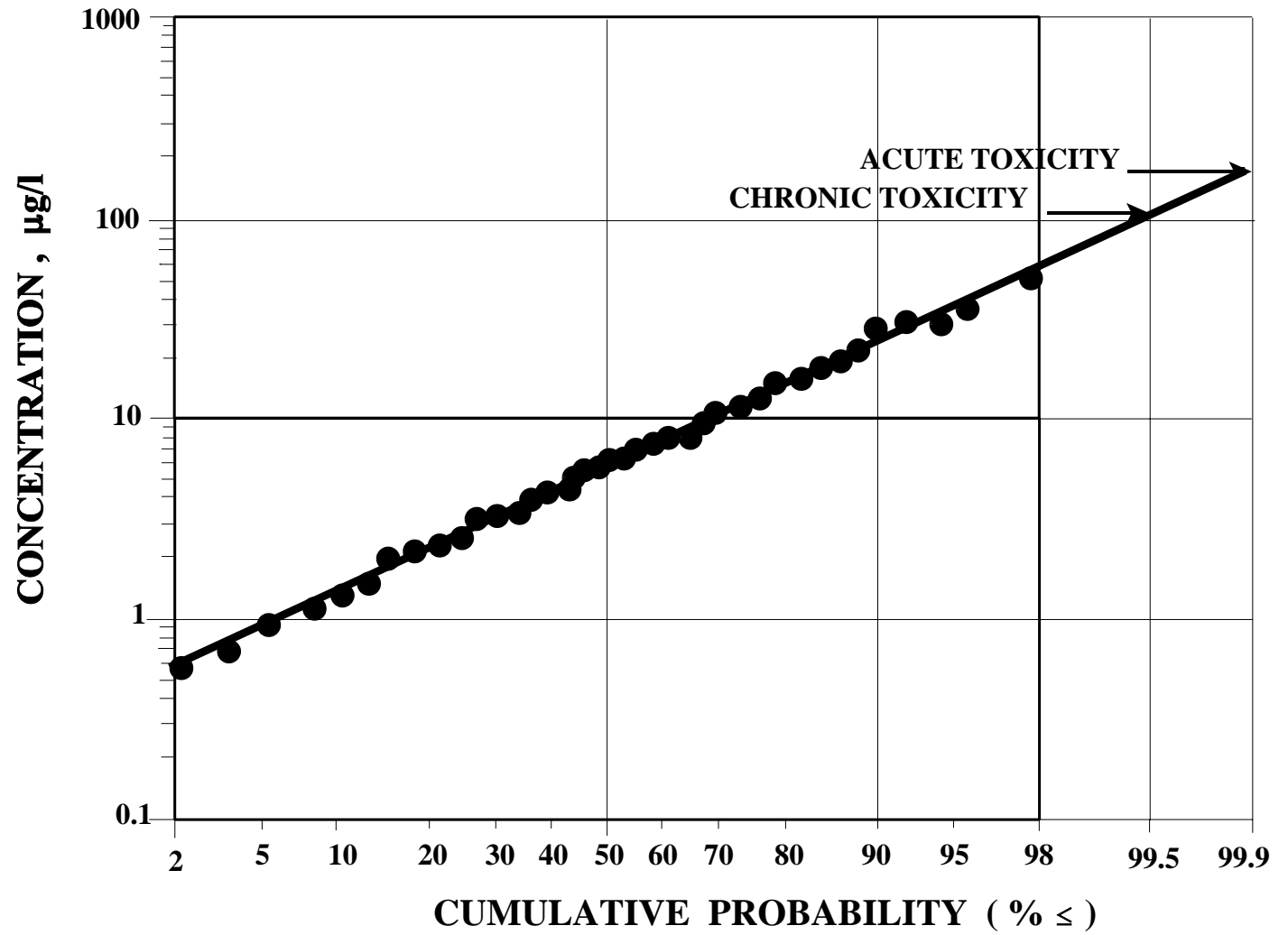
$m_i = W_s / Q$ = average concentration of suspended sediment in the input

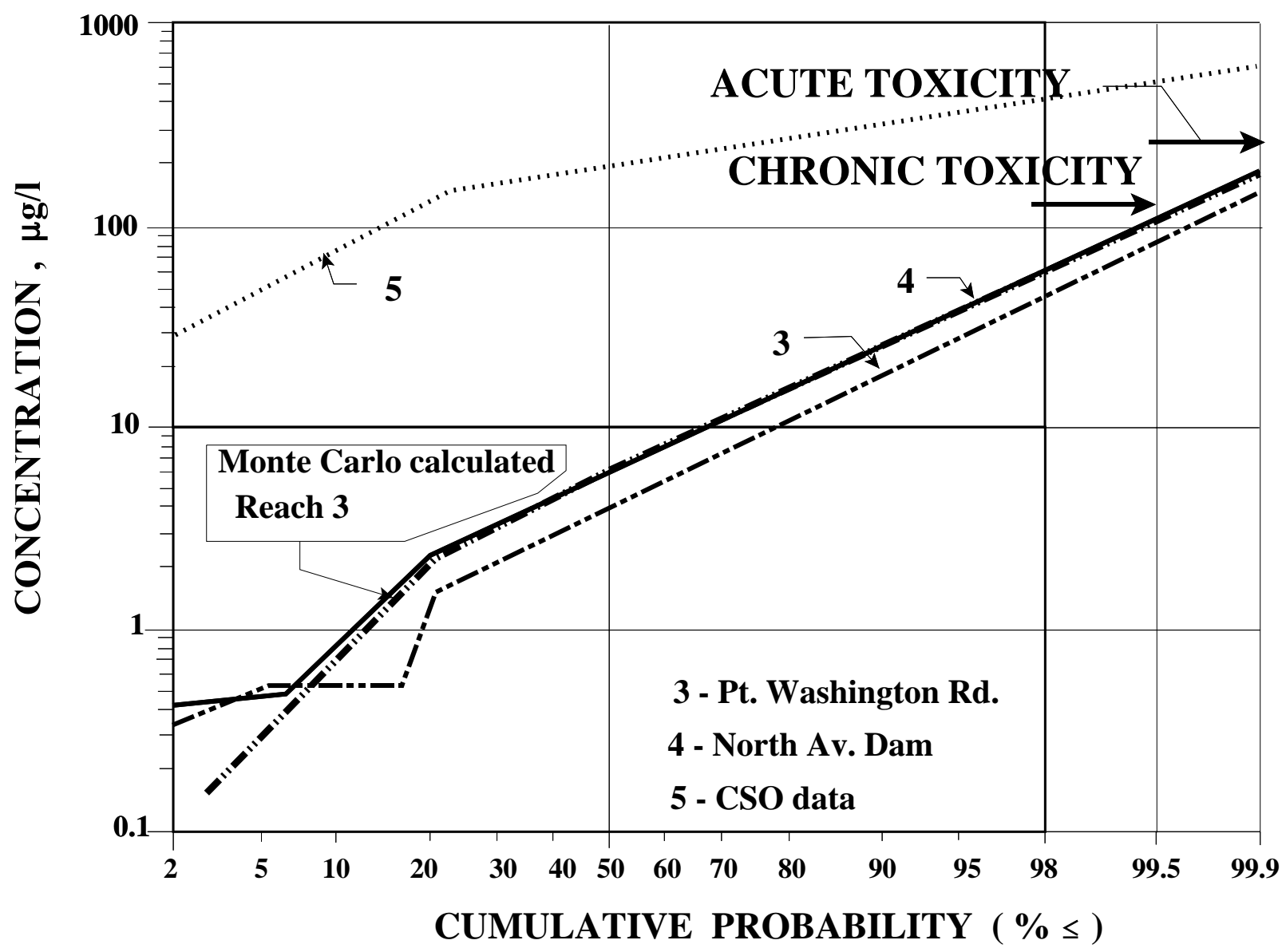
W_s = mass input of suspended solids into the segment

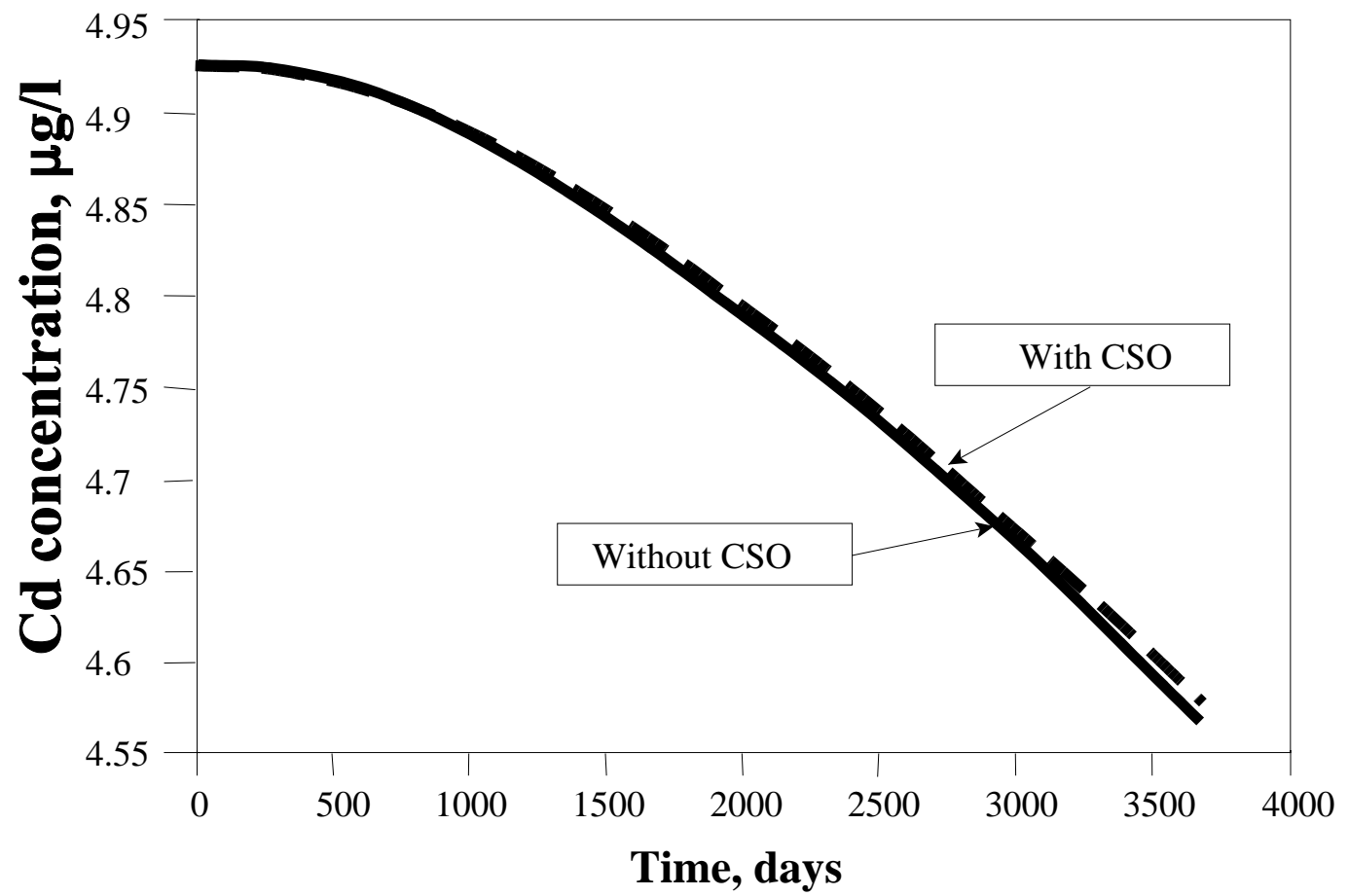
t_0 = residence time in the control volume

K_s = settling velocity or scour rate

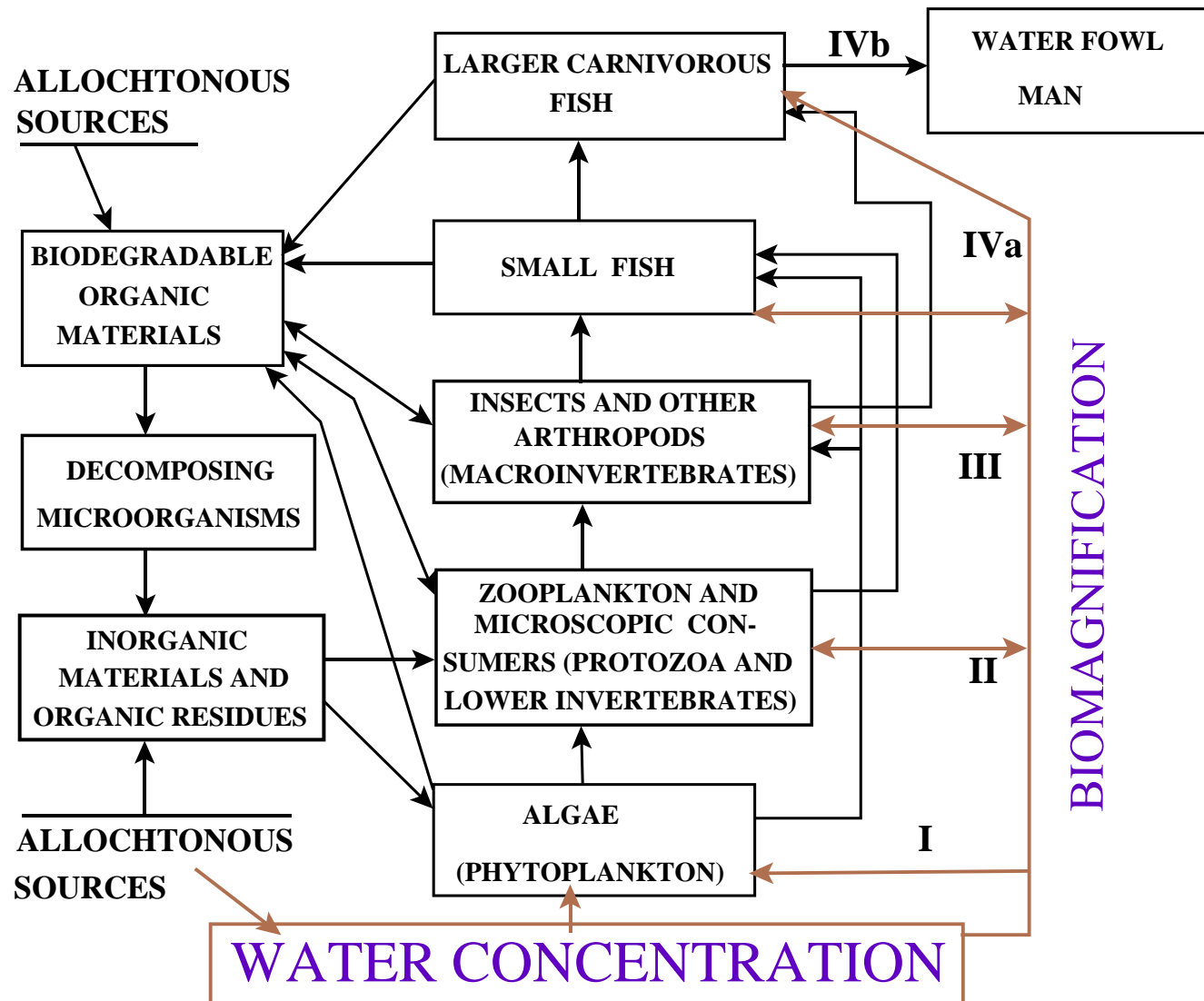








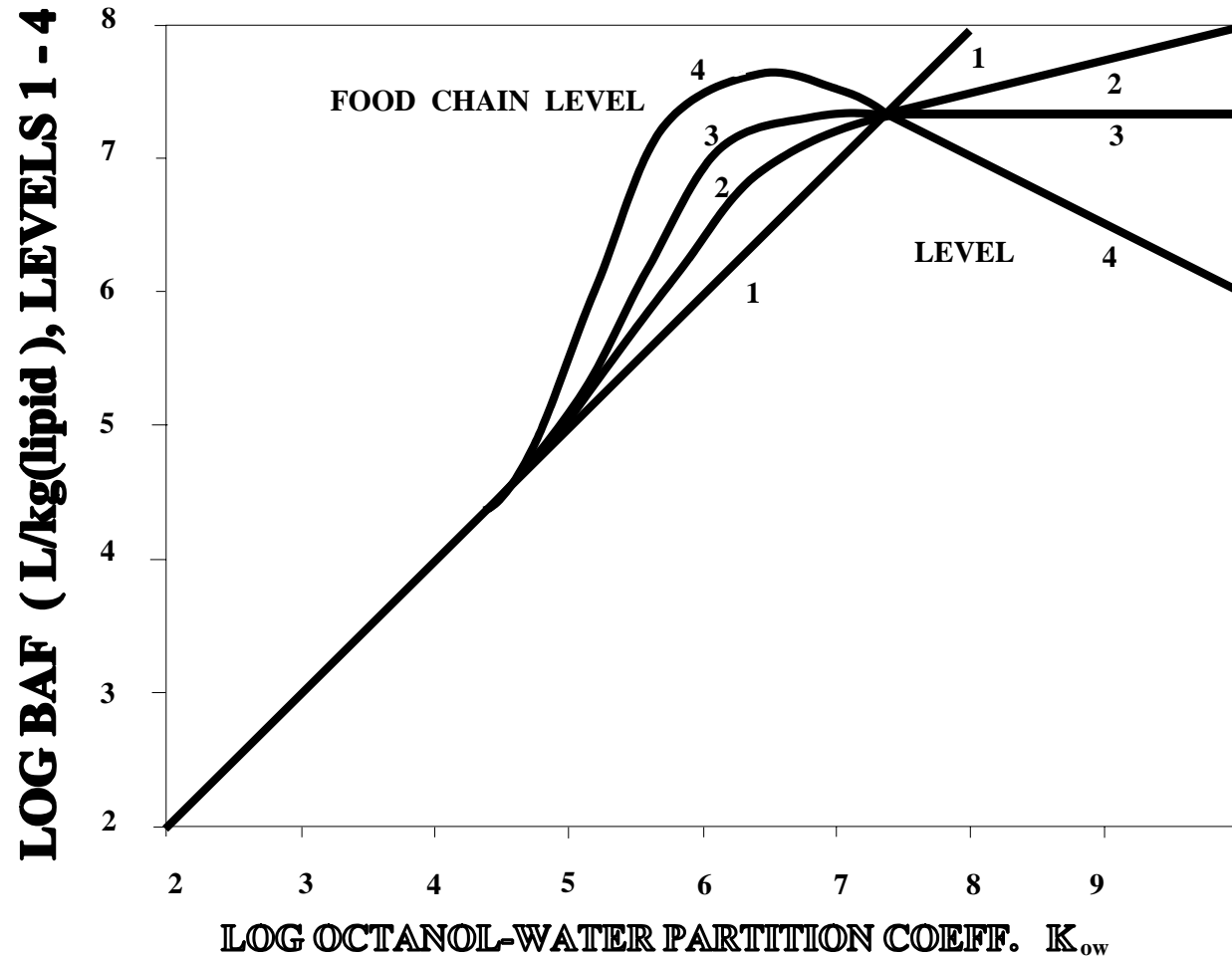
FOOD WEB PROPAGATION



BIOMAGNIFICATION

Concentrations of the toxic compounds increase in the fatty tissues (lipid) of organisms with each trophic level

Body buildup = intake - euration



$$C(\mu\text{g/g of lipid}) = \text{BAF} \times C(\mu\text{g/L in water})$$

MODELS CAPABLE TO SIMULATE FATE OF TOXICS

DYNTOX	River	Far field	Organics, metals
WASP4	Lake, river Estuary	Far field	Organics, metals
HSP-F	River	Far field	Organics, metals
MINTEQUA2	Lake,river Estuary		Metals